

OPACITIES: LTE & NON-LTE

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 Methods in Plasma Spectroscopy



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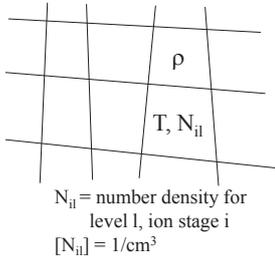
Acknowledgements

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 Honglin Zhang

Some helpful illustrations

Our sample plasma made up of cells described by temperatures, densities, atomic populations, etc.



Our sample ions/atoms inhabiting each cell

level l	ion i	ion (i+1)	ion (i+2)
∞	.	.	.
.	.	.	.
.	.	.	.
3	—	—	—
2	—	—	—
1	—	—	—
(e.g.)	neutral i=1	singly ionized i=2	doubly ionized i=3

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The main players

- Photons (radiation field)
- Bound electrons (orbiting around the nucleus)
- Free electrons (formerly bound electrons that have been ionized by free electrons or photons)
- The photons and electrons interact via fundamental atomic processes, which can be used to determine the atomic level populations, N_{il}
- These populations can then be used to compute an opacity, κ_ν , which is used in radiation transport calculations

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Explanatory definitions/symbols

- ρ - ion (or material) mass density: $[\rho] = \text{grams/cm}^3$
- $N_i = \sum_l N_{il}$ - ion number density: $[N_i] = 1/\text{cm}^3$; $\{N_i = \rho(A_0/A)\}$
- N_e - free electron number density: $[N_e] = 1/\text{cm}^3$
- T or kT = temperature (ion, electron, radiation): $[T] = \text{eV}$
- $\bar{Z}(\rho, T)$ ("Z bar") or $\langle Z \rangle$ - average charge state ($N_e = \bar{Z}N_i$)
- $h\nu$ - photon energy: $[h\nu] = \text{eV}$
- $\kappa_\nu(\rho, T)$ - opacity: $[\kappa_\nu] = \text{cm}^2/\text{gram}$
- $\epsilon_\nu(\rho, T)$ - emissivity: $[\epsilon_\nu] = \text{ergs}/(\text{gram sec Hz})$
- I_ν - (isotropic) radiation intensity: $[I_\nu] = \text{ergs}/(\text{cm}^2 \text{sec Hz})$

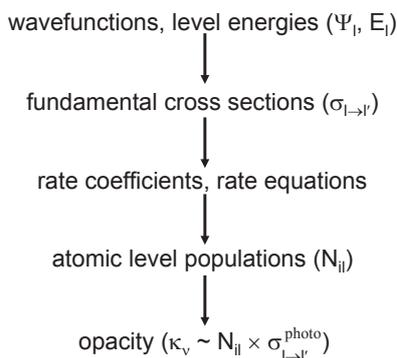
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Atomic kinetics modeling is an *ab-initio* effort

- There are far too many atomic processes to be measured experimentally
- Furthermore, there are not many experimental measurements of atomic physics data
- Nuclear data are obtained through evaluations which rely on both experimental data and theoretical calculations
- Atomic data (e.g. opacities) are obtained almost exclusively from first-principle calculations (quantum mechanics, wavefunctions, cross sections, etc.)

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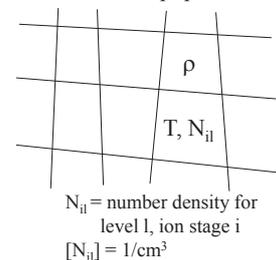
Road map to opacity



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Excitation and de-excitation processes

Our sample plasma made up of cells described by temperatures, densities, atomic populations, etc.



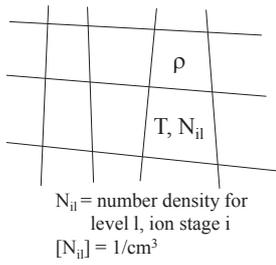
Our sample ions/atoms inhabiting each cell

level l	ion i	ion (i+1)	ion (i+2)
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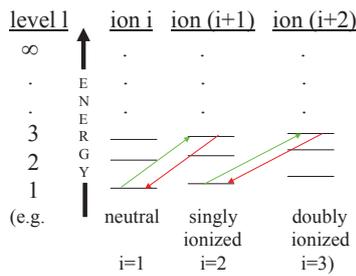
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Ionization and recombination processes

Our sample plasma made up of cells described by temperatures, densities, atomic populations, etc.



Our sample ions/atoms inhabiting each cell



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Solving for the atomic level populations, N_{il}

- To obtain an opacity at each point in our sample plasma, we require the fundamental cross sections and the level populations, N_{il}
- The level populations are determined by the following basic atomic processes and their inverses:

process	inverse process
photoexcitation	photo de-excitation
photoionization	radiative recombination
electron collisional excitation	electron collisional de-excitation
electron collisional ionization	three-body recombination
autoionization	dielectronic recombination
- The cross sections for these processes are used in coupled, differential equations, known as "rate equations", which determine the populations N_{il}

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The rate equations

- In general, the level populations vary as a function of time
- One must consider all possible processes that can populate and depopulate each level
- The result is a set of non-linear, first-order differential equations
- $\frac{dN_{il}}{dt} = (\text{Formation rates}) - (\text{Destruction rates})$

In matrix form

$$\begin{pmatrix} \frac{dN_{i1}}{dt} \\ \dots \\ \frac{dN_{il}}{dt} \\ \dots \\ \frac{dN_{im}}{dt} \end{pmatrix} = \begin{pmatrix} R_{i1} & \dots & R_{im} \\ \dots & \dots & \dots \\ R_{l1} & \dots & R_{lm} \\ \dots & \dots & \dots \\ R_{m1} & \dots & R_{mm} \end{pmatrix} \begin{pmatrix} N_{i1} \\ \dots \\ N_{il} \\ \dots \\ N_{im} \end{pmatrix}$$

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The rate equations (continued)

- The order of the rate matrix can vary greatly depending on the complexity of the atomic model
- Average-atom: order ~ 10 , very crude, very fast to compute
- Configuration-average: order $\sim 100-10^7$, good compromise, some spectral detail, but maybe not enough to produce high-resolution spectra
- Fine-structure: order $\sim 100-10^{10}$, spectrally resolved features, very accurate if complete model can be considered, but can be impractical to solve numerically

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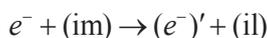
A specific example: collisional excitation/de-excitation

- Each element of the rate matrix is computed from fundamental cross sections associated with each process
- Consider collisional excitation/de-excitation as a specific example:

$$\frac{dN_{il}}{dt} = (\text{"rate" of excitations into il}) - (\text{"rate" of de-excitations out of il}) + \dots$$

$$= \sum_m [s(\text{im,il};T)N_e N_{im} - t(\text{im,il};T)N_e N_{il}] + \dots$$

- $s(\text{im,il};T)$ is the "rate coefficient" for electron collisional excitation from level m to level l in ion stage i , symbolically written as



- Similarly, $t(\text{im,il};T)$ represents the rate coefficient for all possible collisional de-excitations into level l of ion stage i

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A specific example: collisional excitation/de-excitation (continued)

- The result looks like

$$s(\text{im,il};T_e) = \int_{E_0}^{\infty} F(E, T_e) v(E) \sigma_{\text{iml}}(E) dE$$

- $F(E, T_e)$ is the free-electron energy distribution function
- $v(E)$ is the velocity of a free electron [$v(E) = \sqrt{(2E)/m_e}$]
- $\sigma_{\text{iml}}(E)$ is the excitation cross section
- E_0 is the threshold energy, above which excitation can occur
- The **rate** at which excitations occur from level m to level l is $s(\text{im,il};T)N_e$ and the **rate per unit volume** is $s(\text{im,il};T)N_e N_{im}$

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A specific example: collisional excitation/de-excitation (continued)

- The rate coefficients for collisional de-excitation are determined from the principle of detailed balance and can also be expressed in terms of the same excitation cross section
- The rate coefficients for the remaining collisional and photo processes are determined in a similar fashion
- Just as electron-collision processes require a knowledge of the free-electron energy distribution function, $F(E, T_e)$, photo processes require that the photon energy distribution function also be specified
- These concepts lead naturally to a discussion of LTE vs. non-LTE (NLTE) atomic physics

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Free electrons in thermodynamic equilibrium

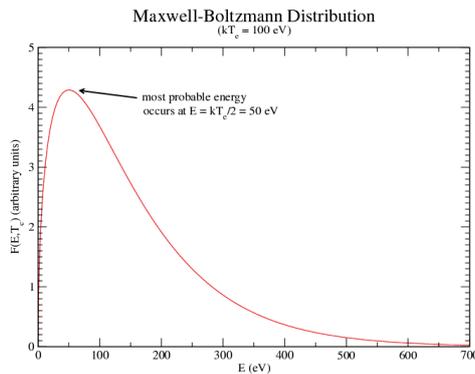
- If the free electrons are in thermodynamic equilibrium (TE) with themselves, then the energy distribution is given by the Maxwell-Boltzmann distribution at an electron temperature T_e

$$F(E, T_e) = \frac{2}{\sqrt{\pi}} \frac{\sqrt{E}}{(kT_e)^{3/2}} e^{-E/kT_e}$$

- This distribution represents the fraction of electrons per unit energy interval that have energies between E and $E+dE$

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Maxwellian distribution at $kT_e=100$ eV



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Photons in thermodynamic equilibrium

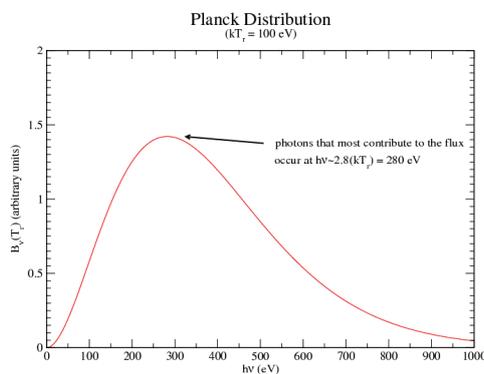
- Similarly, if the photons are in thermodynamic equilibrium (TE) with themselves, then the energy density distribution is given by the Planck distribution at a radiation temperature T_r

$$B_\nu(T_r) = \frac{2}{(hc)^2} \frac{(h\nu)^3}{e^{h\nu/kT_r} - 1}$$

- This is a flux distribution that represents the amount of radiation energy per unit frequency interval per unit area per unit time per unit solid angle

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Planckian distribution at $kT_r=100$ eV



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Local Thermodynamic Equilibrium

- LTE = Local Thermodynamic Equilibrium
- LTE is valid at a particular point in the plasma if the electron and photon distributions are in equilibrium **with each other**: $T_e = T_r = T_i = T$. This is one of the “textbook” definitions of LTE.
- There are other descriptions of LTE...

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LTE applies if any of the following are true:

- At a given point in the plasma, the (atomic) conditions can be described by a single temperature ($T = T_e = T_r = T_i$)
- The rate at which any atomic process occurs is exactly balanced by the rate of its inverse process (this condition makes the physics much simpler to deal with than NLTE)
- The energy distribution of the free electrons in the plasma is described by a Maxwellian distribution and the radiation field is described by a Planck function (all at the same temperature)
- The FREE ELECTRON density is so high that electron collisions dominate the various atomic processes (“collision-dominated plasma”). In this case, there is not a true balance between all processes, but the following, and perhaps most important, bullet is still true:

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LTE from a practical (computational) perspective

- From a computational perspective, LTE means that the atomic level populations, N_{il} , can be solved from the (relatively) simple Saha equation and the Boltzmann relationship

$$N_{il} \propto (N_i) e^{-E_{il}/kT}$$

- In this case, the N_{il} can be determined from a simple analytic formula that depends on the energy and temperature; there is **no need to consider the fundamental cross sections**.
- Solving the detailed rate equations with a Maxwellian electron distribution and a Planckian radiation distribution results in a steady-state solution ($dN_{il}/dt = 0$) which could have been found by solving the much simpler Boltzmann relationship above

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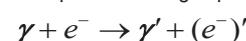
Non-LTE

- Non-LTE applies if:
 - LTE conditions are not satisfied (obviously!)
 - System is changing so rapidly that electron and/or photon energy distributions do not reach thermal equilibrium (i.e. Maxwellian or Planckian is not valid, lasers, $T_r \neq T_e$, etc.)
 - Optically thin plasma: radiation escapes and is not available to provide LTE balance among the fundamental atomic processes
- For the NLTE case, the detailed rate equations must be solved to obtain the atomic level populations, N_{il}
- In practice, this solution requires the use of large-scale computing
- NLTE calculations can take as much as 3-4 **orders of magnitude** more computing time than LTE calculations

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Photon scattering

- One additional, fundamental process must be discussed before an opacity can be constructed: Compton scattering of photons



- This process differs from free-free absorption in that the incident photon loses only a small portion of its energy when interacting with a free electron, then continues on with a slightly smaller energy

$$\sigma_{\text{THOMSON}} = \frac{8\pi}{3} \left(\frac{e^2}{m_e c^2} \right)^2 = 6.66 \times 10^{-25} \text{ (cm}^2\text{)} \quad (h\nu \ll mc^2)$$

$$\sigma_{\text{COMPTON}}(\nu) = G(\nu) \sigma_{\text{THOMSON}}$$

- $G(\nu)$ is a relativistic correction factor that accounts for the case when the photon energy becomes comparable to the electron rest mass and the electron's kinetic energy is treated in a fully relativistic manner

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What is an opacity?

- An opacity, κ_ν , describes the coupling between matter and radiation via electron-photon interactions
- Opacity gives a measure of how much radiation a certain material will absorb/scatter (i.e. how “opaque” is the material)
- An opacity can be thought of as a macroscopic quantity that is built up from fundamental atomic cross sections
- The amount of radiation that is absorbed/scattered (i.e. removed) from the ambient radiation field, I_ν , in each cell of our sample plasma is given by:

$$\kappa_\nu I_\nu$$

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What is an emissivity?

- An emissivity, ϵ_ν , gives the amount of radiation that will be emitted by the material in a plasma via electron-photon interactions
- As with the opacity, an emissivity is calculated from fundamental atomic cross sections
- The amount of radiation that is emitted (i.e. added to) the ambient radiation field, I_ν , in each cell of our sample plasma is given by:

$$\epsilon_\nu / (4\pi) \quad (\text{isotropic emitter})$$

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Why are opacities/emissivities important?

- These quantities are necessary to solve the radiation transport equation
- Assuming problem is time-independent and one-dimensional with isotropic radiation, the transport equation can be written:

$$\frac{1}{\rho} \frac{dI_\nu}{dx} = \frac{\epsilon_\nu}{4\pi} - \kappa_\nu I_\nu$$

Labels in diagram: material density (ρ), emissivity (ϵ_ν), radiation intensity (I_ν), opacity (κ_ν), radiation frequency (ν).

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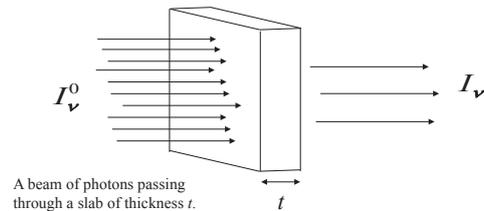
The classic opacity (transmission) experiment: Optically thin plasma example

- If the plasma is “optically thin”, then the emitted radiation will escape and need not be considered in the radiation transport equation:

$$\frac{1}{\rho} \frac{dI_\nu}{dx} = \frac{\epsilon_\nu}{4\pi} - \kappa_\nu I_\nu$$

(Note: The $\frac{\epsilon_\nu}{4\pi}$ term is crossed out with a red X in the original image.)

- This situation can be illustrated by the following diagram:



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Optically thin plasma example (continued)

- The previous differential equation has a well-known solution:

$$I_\nu = I_\nu^0 e^{-(\rho \kappa_\nu t)}$$

- This sort of “transmission experiment” is the typical way in which opacities are measured
- The quantity $\lambda_\nu^{\text{mfp}} = (1/\rho \kappa_\nu)$ has the dimensions of length and is called the **optical mean free path**. The mean free path is a useful physical quantity and is defined as the average distance a photon can travel through a material without being absorbed or scattered. Optically thin plasmas have physical dimensions $\ll \lambda_\nu^{\text{mfp}}$.

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Computing an opacity from fundamental atomic cross sections

- Basically, opacity = (atomic population)(cross section)/(mass density) (NB: we are only interested in **photo** cross sections now)
- When interacting with electrons, a photon can be absorbed (most/all energy given to electrons) or scattered (some energy given to electrons, but photon survives with slightly decreased energy)

$$\kappa_\nu^{\text{TOT}}(\rho, T_e, T_r) = \kappa_\nu^{\text{ABS}}(\rho, T_e, T_r) + \kappa_\nu^{\text{SCAT}}(\rho, T_e, T_r)$$

$$\kappa_\nu^{\text{ABS}} = \frac{1}{\rho} \sum_{ii} N_{ii}(\rho, T_e, T_r) [\sigma_{ii}^{\text{(bound-bound)}}(\nu) + \sigma_{ii}^{\text{(bound-free)}}(\nu)] + \kappa_\nu^{\text{(free-free)}}$$

Labels in diagram: material density (ρ), atomic level populations (N_{ii}), photoexcitation cross sections ($\sigma_{ii}^{\text{(bound-bound)}}$), photoionization cross sections ($\sigma_{ii}^{\text{(bound-free)}}$), Compton scattering (κ_ν^{SCAT}), inverse Bremsstrahlung contribution ($\kappa_\nu^{\text{(free-free)}}$).

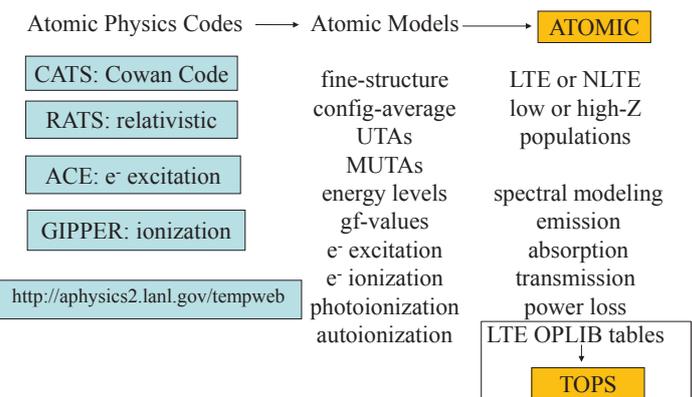
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How to compute an opacity

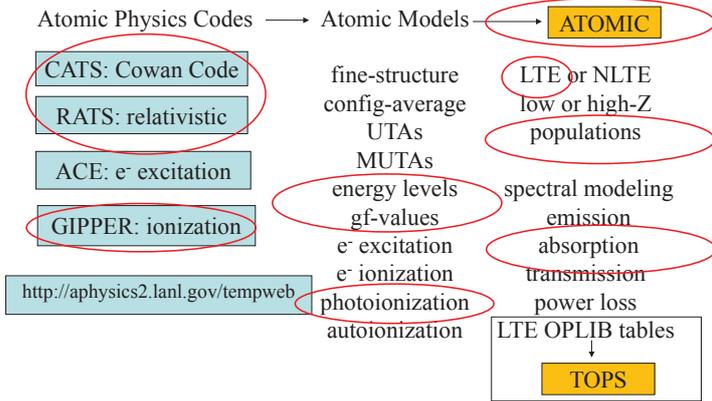
- Compton scattering uses a straightforward formula: $\kappa_\nu^{\text{SCAT}} = N_e \sigma^{\text{SCAT}}(\nu) / \rho$ [$\approx 0.4Z/A$ (cm^2/g) for Thomson scattering]
- The free-free contribution is straightforward (Kramers’ formula)
- The bound-bound and bound-free contributions are obtained by summing over ALL bound levels of ALL important ion stages
- This sum requires the populations, N_{ii} , as well as the relevant photo cross sections, $\sigma_{ii}^{\text{photo}}$
- The previous opacity equations are valid for both LTE and NLTE conditions
- The LTE/NLTE difference **is in how one calculates the atomic populations**, N_{ii}

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The LANL Suite of Atomic Modeling Codes

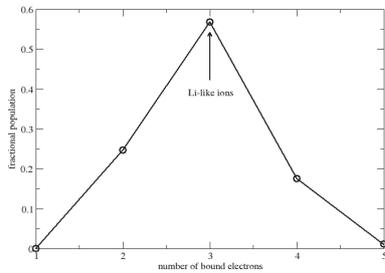


To calculate LTE opacities, you need only:



Numerical example of an LTE opacity: Aluminum plasma at kT = 40 eV, N_e = 10¹⁹cm⁻³

- For these conditions, $\langle Z \rangle = 10.05 \Rightarrow$ there is an average of ~2.95 bound electrons/ion (Li-like ions are dominant)
- Here is the charge state distribution:

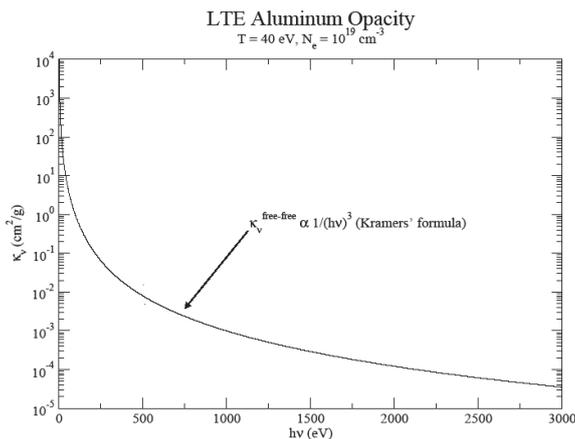


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Numerical example of an LTE opacity: Aluminum plasma at kT = 40 eV, N_e = 10¹⁹cm⁻³

- For these conditions, $\langle Z \rangle = 10.05 \Rightarrow$ there is an average of ~2.95 bound electrons/ion (Li-like ions are dominant)
- The following plots show the contribution to the total opacity from each of the three photo-absorption processes as well as the contribution from Compton scattering
- You will see some arcane spectroscopic notation: bound electrons with the same principal quantum number n are said to inhabit the same "shell". Each shell is identified by a capital letter: $n=1$, K-shell $n=2$, L-shell $n=3$, M-shell
- Bound-bound absorption involving an active bound electron that initiates from the K-shell is referred to as "K-shell" absorption, etc. Bound-bound emission that terminates with a bound electron ending up in the K-shell is referred to as "K-band" emission, etc.

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What about emissivities?

- Simple relationship for LTE conditions:

$$\text{emissivity} \rightarrow \epsilon_\nu = (4\pi)\kappa_\nu^{\text{ABS}}(\rho, T)B_\nu(T)$$

Planck function

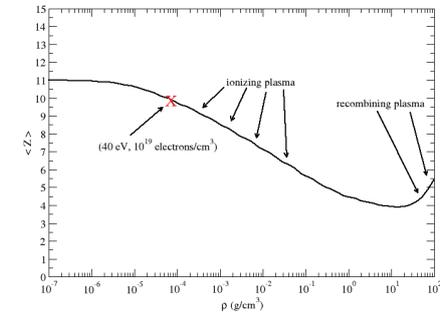
opacity

- One only needs the opacity to obtain the emissivity when doing LTE calculations
- Non-LTE emissivities require the level populations, N_{ij} , along with the cross sections for the *inverse* of the photo-absorption processes that were considered for opacities

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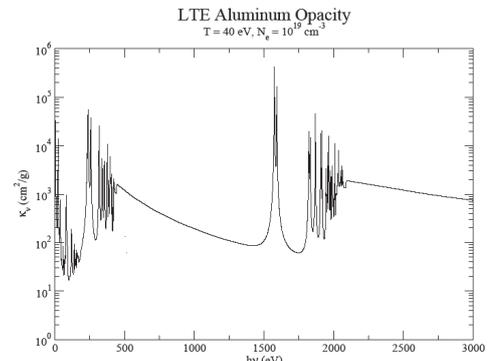
Another useful plot to consider: $\langle Z \rangle$ vs. ρ

- Here is a plot of $\langle Z \rangle$ vs mass density for a fixed temperature of 40 eV:

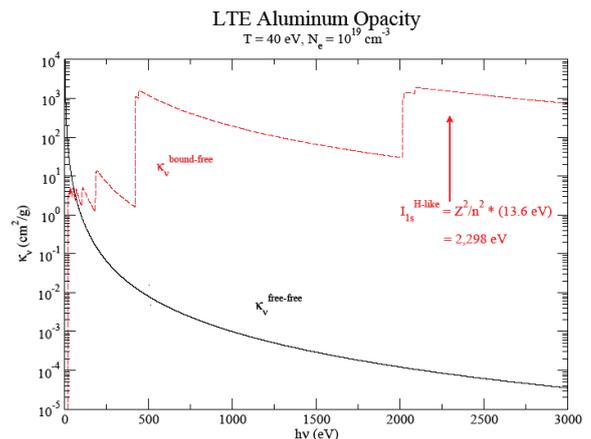


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First, a snapshot of the total LTE opacity for this aluminum plasma



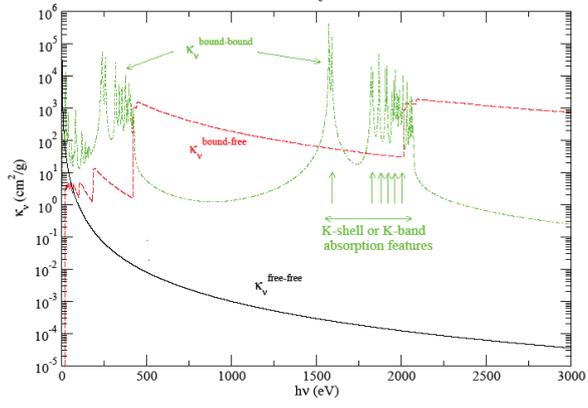
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LTE Aluminum Opacity

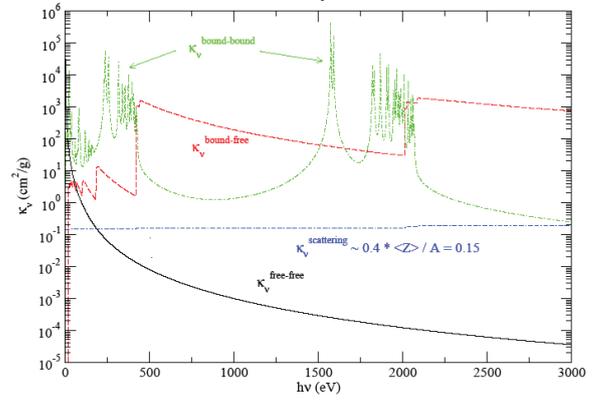
$T = 40 \text{ eV}, N_e = 10^{19} \text{ cm}^{-3}$



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LTE Aluminum Opacity

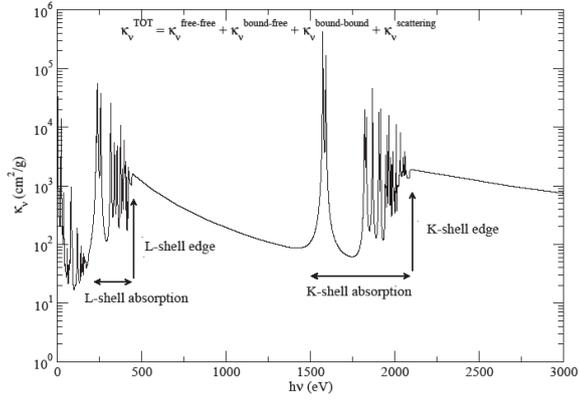
$T = 40 \text{ eV}, N_e = 10^{19} \text{ cm}^{-3}$



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LTE Aluminum Opacity

$T = 40 \text{ eV}, N_e = 10^{19} \text{ cm}^{-3}$



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